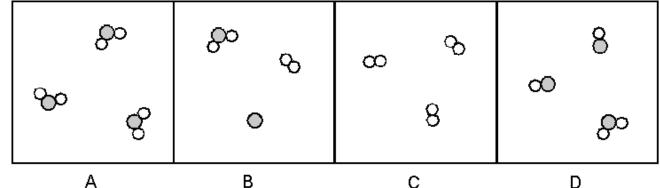
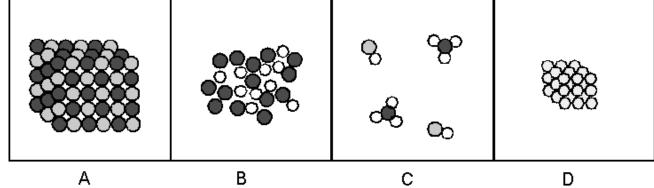
## **Unit 6 – Exercise 6 Substance vs Mixture**

1.	Describe a way to separate a mixture of sand and salt?
2.	Describe a way to separate a mixture of salt and water?
3.	Explain why using a filter would not be effective in separating a mixture of salt and sugar.
4.	Draw particle representations for the following:
	A mixture of iron and sulfur A compound of iron and sulfur
5.	Explain why a magnet can separate iron particles from the mixture but not from the compound.

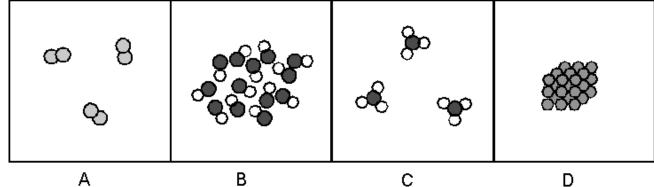
- 6. Based on the particle diagram given in each box for questions 7, 8 & 9 label the material as a solid, liquid or gas
- 7. Consider the four containers below.



- a. Which of these are mixtures? \_\_\_\_\_ pure substances? \_\_\_\_\_
- b. Which contain only compounds? \_\_\_\_\_ only elements \_\_\_\_
- 8. Consider the four containers below.



- a. Which of these are mixtures? \_\_\_\_\_ pure substances? \_\_\_\_\_
- b. Which contain only compounds? \_\_\_\_\_ only elements \_\_\_\_\_
- 9. Consider the four containers below.



- a. Which of these are mixtures? \_\_\_\_\_ pure substances? \_\_\_\_\_
- b. Which contain only compounds? \_\_\_\_\_ only elements \_\_\_\_\_