# **Physical Science Reference Tables**

#### **MOTION AND ENERGY**

$$\overline{v} = \frac{\Delta d}{\Delta t}$$

$$a = \frac{v_f - v_i}{\Delta t}$$

$$F = ma$$

$$F_a = mg$$

$$p = mv$$

$$W = F\Delta d$$

$$P = \frac{W}{\Delta t}$$

$$PE_g = mgh = F_gh$$

$$KE = \frac{1}{2}mv^2$$

$$IMA = \frac{d_E}{d_R}$$

$$AMA = \frac{F_R}{F_E}$$

Efficiency = 
$$\frac{W_{out}}{W_{in}} \times 100$$

$$V_w = f\lambda$$

$$v = velocity$$

$$d = position$$

$$t = time$$

$$a = uniform acceleration$$

$$F = force$$

$$m = mass$$

$$F_g$$
 = weight

$$g$$
 = acceleration due to gravity on Earth

$$= 9.8 \text{ m/s/s}$$

$$p = momentum$$

$$W = work$$

$$P = power$$

$$PE_g$$
 = gravitational potential energy

$$h = height$$

$$R = resistance$$

$$E = effort$$

$$v_w$$
 = wave velocity

$$f = frequency$$

$$\lambda = wavelength$$

#### **ELECTRICITY**

$$V = IR$$

$$P = VI$$

$$V = IR$$

$$P = V1$$

$$D = \frac{m}{V}$$

$$m = mass$$

P = power

I = current

R = resistance

V = electrical potential difference

$$V = \text{volume}$$

## **PERIODIC TABLE**

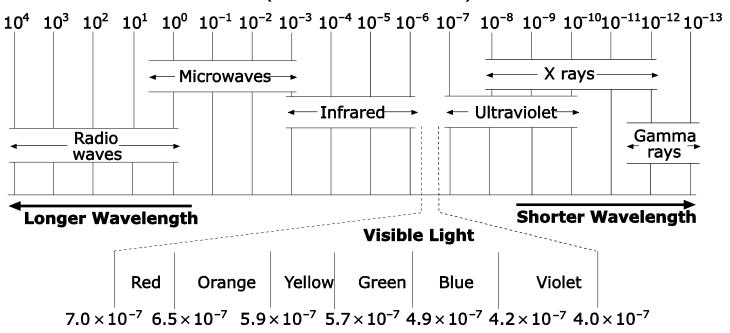
1 IA								
1 H Hydrogen 1.008	2 IIA							
3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012							
11 <b>Na</b> Sodium 22.99	12 <b>Mg</b> <sub>Magnesium</sub> 24.31	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIIIB	9 VIIIB
19 K Potassium 39.10	20 <b>Ca</b> Calcium 40.08	21 <b>Sc</b> Scandium 44.96	22 <b>Ti</b> Titanium 47.88	23 <b>V</b> Vanadium 50.94	24 Cr Chromium 51.99	25 <b>Mn</b> Manganese 54.94	26 <b>Fe</b> Iron 55.85	27 <b>Co</b> Cobalt 58.93
37 <b>Rb</b> Rubidium 85.47	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.91	40 <b>Zr</b> Zirconium 91.22	41 <b>Nb</b> Niobium 92.91	42 <b>Mo</b> Molybdenum 95.94	43 Tc Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.91
55 <b>Cs</b> Cesium 132.91	56 <b>Ba</b> Barium 137.38	57 <b>La</b> Lanthanum 138.91	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.95	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.21	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.22
87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium (226)	89 Ac Actinium (227)	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 Sg Seaborglum (263)	107 <b>Bh</b> Bohrium (264)	108 <b>Hs</b> Hassium (269)	109 Mt Meitnerium (268)
		58 <b>Ce</b> Cerium 140.12	59 <b>Pr</b> Praseodymium 140.91	60 <b>Nd</b> Neodymium 144.24	61 Pm Promethium (145)	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.96	64 <b>Gd</b> Gadolinium 157.25
		90 <b>Th</b> Thorium 232.04	91 <b>Pa</b> Protactinium 231.04	92 <b>U</b> Uranium 238.04	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)

### **OF THE ELEMENTS**

								18
								VIIIA
								2
			13	14	15	16	17	He Helium
			IIIA	IVA	VA	VIA	VIIA	4.003
			5	6	7	8	9	10
			<b>B</b> Boron	<b>C</b> Carbon	<b>N</b> Nitrogen	Oxygen	<b>F</b> Fluorine	Ne Neon
			10.81	12.01	14.01	16.00	19.00	20.18
			13	14	15	16	17	18
10	11	12	<b>Al</b> Aluminum	<b>Si</b> Silicon	P Phosphorus	<b>S</b> Sulfur	<b>CI</b> Chlorine	<b>Ar</b> Argon
VIIIB	IB	IIB	26.98	28.09	30.97	32.07	35.45	39.95
28	29	30	31	32	33	34	35	36
<b>Ni</b> Nickel	Cu	<b>Zn</b> Zinc	<b>Ga</b> Gallium	Ge	As	Se	Br	Kr
58.69	Copper 63.55	65.39	69.72	Germanium 72.61	Arsenic 74.92	Selenium 78.96	Bromine 79.90	Krypton 83.80
46	47	48	49	50	51	52	53	54
Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Palladium 106.42	Silver 107.87	Cadmium <b>112.41</b>	Indium 114.82	Tin 118.71	Antimony 121.76	Tellurium 127.60	Iodine 126.90	Xenon 131.29
78	79	80	81	82	83	84	85	86
Pt	Au	Hg	TI	Pb	Bi	Po	<b>At</b> Astatine	<b>Rn</b> Radon
Platinum 195.08	Gold 196.97	Mercury 200.59	Thallium 204.38	Lead 207.20	Bismuth 208.98	Polonium (209)	(210)	(222)
110	111	112						
Ds	Rg	Cn						
Darmstadtium (271)	Roentgenium (272)	Copernicium (285)						
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65	66	67	68	69	70	71
<b>Tb</b>	<b>Dy</b>	<b>Ho</b>	<b>Er</b>	<b>Tm</b>	<b>Yb</b>	<b>Lu</b>
Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium
158.93	162.50	164.93	167.26	168.93	173.04	174.97
97	98	99	100	101	102	103
<b>Bk</b>	<b>Cf</b>	Es	<b>Fm</b>	Md	<b>No</b>	Lr
Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium
(247)	(251)	(252)	(257)	(258)	(254)	(262)

## **Electromagnetic Spectrum** (measurement in meters)



Polyatomic Ions				
NH <sub>4</sub>	Ammonium			
C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> <sup>-</sup>	Acetate			
CIO <sub>3</sub>	Chlorate			
NO <sub>3</sub>	Nitrate			
OH <sup>-</sup>	Hydroxide			
CO <sub>3</sub> <sup>2-</sup>	Carbonate			
SO <sub>4</sub> <sup>2-</sup>	Sulfate			
PO <sub>4</sub> <sup>3-</sup>	Phosphate			